

Fig. 5. Projection along the b -axis showing the methyl contact region. Non-equivalent hydrogen-hydrogen distances (in Å) less than 3.0 Å over the methyl gap are given.

The hydrogen-hydrogen contact distances over the methyl gap (Fig. 5) give a more informative picture of the packing of the methyl-group planes than the carbon-carbon distances discussed. The shortest distance is 2.62 Å, somewhat longer than the shortest lateral hydrogen-hydrogen contact between the chains, which is about 2.40 Å.

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The Crystal Structure of Monoaquobisacetylacetonatozinc

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The crystal structure of monoaquobisacetylacetonatozinc [$\text{Zn}(\text{C}_5\text{H}_7\text{O}_2)_2\text{H}_2\text{O}$] has been determined by three-dimensional methods. The cell dimensions are $a = 10.48$, $b = 5.37$, $c = 10.94$ Å, $\beta = 93^\circ 48'$, space group $P2_1$, with $Z = 2$. The structure is made up of discrete molecules, containing 5-coordinate zinc with a coordination configuration intermediate between tetragonal pyramidal and trigonal bipyramidal.

Introduction

Lippert & Truter (1960) have recently published a determination of the crystal structure of mono-aquobisacetylacetonatozinc, $\text{Zn}(\text{C}_5\text{H}_7\text{O}_2)_2 \cdot \text{H}_2\text{O}$. Since

we had an independent structure determination in the final refinement stages at the time we learned of their work, and since there appeared to be some significant differences in the results, we have completed our determination.

Experimental

Monoaquobisacetylacetonatozinc crystallized from benzene, ethanol or benzene-ethanol mixtures in needles. The unit cell dimensions were determined from rotation and zero level Weissenberg photographs about the needle (b) axis using $\text{Cu } K\alpha$ radiation with superimposed reflections from a sodium chloride crystal ($a_0=5.6387 \text{ \AA}$) for calibration. In addition, zero, first and second level Weissenberg exposures and zero level precession photographs ($\text{Mo } K\alpha$, $\lambda=0.7107 \text{ \AA}$) were taken to establish the space group. The results were:

$$a_0 = 10.48_0 \pm 0.01, \quad b_0 = 5.37_0 \pm 0.005, \\ c_0 = 10.93_5 \pm 0.01 \text{ \AA}, \quad \beta = 93^\circ 48' \pm 9'.$$

The space group was $P2_1$ or $P2_1/m$ ($0k0$ present only for $k=2n$). Calculated density for two molecules per unit cell 1.522 g.cm^{-3} ; observed density 1.51 g.cm^{-3} . A further set of calibrated rotation exposures about b , using iron radiation ($\lambda=1.9373 \text{ \AA}$), were taken to confirm the value of b_0 ($b_0=5.373 \pm 0.005 \text{ \AA}$).

The intensities for the four levels ($h0l$ to $h3l$) were recorded on Kodak medical X-ray film, using multiple films, in a Nonius integrating Weissenberg camera. Considerable difficulty was found in obtaining a crystal that would last through the series of exposures. The intensities were measured photometrically and corrected for Lorentz and polarization factors but not for absorption.

Determination and refinement of the structure

To carry out a structure determination, a choice had to be made between the two space groups $P2_1$ and $P2_1/m$. After experiments with models, the space group $P2_1$ was selected, since it proved very difficult to fit any reasonable model in the cell in a $P2_1/m$ arrangement. This choice was confirmed by the successful determination of the structure. Calculations were carried out for the $h0l$ projection first, using the heavy atom technique. Because of the short b axis all atoms were clearly resolved and accurate x and z coordinates were obtained. With the aid of these coordinates, the zinc and oxygen y parameters were derived from a three-dimensional Patterson function. The first triple Fourier summation clearly showed all the atoms except one, and the structure determination was carried out by a series of difference ($F_o - F_c$) syntheses until the reliability index R was just below 0.10. In the difference maps anisotropy showed strongly in the zinc and oxygen peaks and a change was made to a least-squares cycle of refinement to allow the use of three-dimensional anisotropic temperature factors for these atoms.

The calculations for Fourier summations were carried out on an IBM type 650 computer, with programs devised by Brown, Lingafelter, Stewart & Jensen (1959). The least-squares refinement program

of Busing & Levy (1959) was run on an IBM type 709 computer. The Hughes weighting scheme was used and $\sum w(|F_o| - |F_c|)^2$ was minimized. Atomic scattering factors for zinc were from Thomas & Umeda (1957) and for the other atoms from Berghuis, Haanappel, Potters, Loopstra, MacGillavry & Veenendaal (1955)

Table 1. *Final atomic parameters*

	x/a	y/b	z/c	B	
Zn	0.0892	0.0000	0.2156	*	
O(1)	0.0757	-0.2717	0.0904	*	
O(2)	0.1961	0.2675	0.1410	*	
O(3)	0.2507	-0.1235	0.3107	*	
O(4)	-0.0144	-0.1198	0.3517	*	
O(5)	-0.610	0.2122	0.1577	*	
C(2)	0.3746	0.4640	0.0583	3.78	
C(3)	0.4678	-0.1889	0.3708	4.03	
C(4)	-0.1584	-0.1246	0.5127	2.98	
C(5)	-0.2346	0.5040	0.1733	3.72	
C(22)	0.3203	0.2755	0.1400	2.54	
C(23)	0.4021	0.1200	0.2125	3.04	
C(33)	0.3667	-0.0596	0.2929	2.50	
C(44)	-0.1139	-0.0261	0.3936	2.23	
C(45)	-0.1822	0.1704	0.3354	2.68	
C(55)	-0.1530	0.2806	0.2246	2.17	
C(23) H(1)	0.5035	0.1810	0.2143	†	
C(2) {	H(2)	0.4100	0.6500	0.0900	†
	H(3)	0.3300	0.5000	-0.0100	†
	H(4)	0.5000	0.4300	0.0400	†
C(3) {	H(5)	0.4600	-0.3750	0.3500	†
	H(6)	0.4300	-0.1600	0.4400	†
	H(7)	0.5500	-0.1500	0.3500	†
C(45) H(8)	-0.2600	0.2560	0.3840	†	
C(5) {	H(9)	-0.2800	0.3800	0.0900	†
	H(10)	-0.1800	0.6700	0.2000	†
	H(11)	-0.3700	0.5240	0.2300	†
C(4) {	H(12)	-0.1800	-0.1400	0.5200	†
	H(13)	-0.1700	-0.3250	0.5600	†
	H(13)	-0.1200	-0.0500	0.5500	†
O(1) {	H(15)	0.0400	-0.2250	0.0100	†
	H(16)	0.1600	-0.4000	0.0800	†

* Anisotropic. See Table 2.

† $B=2.50$ used for all hydrogen atoms.

Table 2. *Final anisotropic temperature factors*
($\times 10^2$)

	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Zn	2.62	3.29	3.08	0.09	0.59	0.03
O(1)	5.02	4.36	2.69	0.16	0.25	0.07
O(2)	2.47	3.11	4.96	-0.10	1.19	0.70
O(3)	2.58	5.19	4.33	-0.35	0.49	0.57
O(4)	2.83	4.82	3.14	0.80	1.01	0.69
O(5)	3.37	3.81	2.82	1.42	0.78	0.68

Hydrogen atoms were located by examining the ΔF maps at distances about 1.05 \AA from the carbon atoms and 0.95 \AA from the water oxygen atom. A self consistent set of atomic parameters was chosen. Hydrogen positions were not refined and no great accuracy can be claimed for them. The final reliability factor R was 0.072 (1060 reflections). The results of the analysis are given in Tables 1, 2, and 3.

Table 3. Comparison of observed and calculated structure factors (x 10)

Table with columns for observed (A, L, F0, Fc, Cos alpha, Sin alpha) and calculated (A, L, F0, Fc, Cos alpha, Sin alpha) structure factors. The table contains multiple rows of data for different reflections, with some cells containing asterisks or other symbols indicating specific conditions or errors.

Table 3 (cont.)

<i>A</i>	<i>k</i>	<i>l</i>	<i>F₀</i>	<i>F_c</i>	<i>cos α</i>	<i>sin α</i>	<i>A</i>	<i>k</i>	<i>l</i>	<i>F₀</i>	<i>F_c</i>	<i>cos α</i>	<i>sin α</i>	<i>A</i>	<i>k</i>	<i>l</i>	<i>F₀</i>	<i>F_c</i>	<i>cos α</i>	<i>sin α</i>
283			291	.9960	.0984		180			166	.9838	.1975		142			72	.9944	.1805	
455			.9962	.2070			121			122	.9965	.0381		160			160	.0043	1.0000	
358			.7815	.9963			94			94	.3039	.9257		200			206	.3118	.9052	
321			.9959	.0096			139			139	.9871	.1825		177			177	.0151	.9783	
354			.9180	.3697			148			192	.0710	.9972		52			53	.2196	.9576	
292			.9960	.0000			305			305	.9972	.2213		120			120	.0893	.9690	
235			.8864	.4360			81			81	.0777	.9964		169			169	.0786	.9577	
196			.201	.9766	.2513		90			94	.9976	.0964		79			79	.3385	.9140	
210			.9969	.0000			126			126	.9962	.0781		125			120	.4048	.9414	
116			.9912	.1235			84			84	.9960	.0905		134			127	.3470	.9587	
73			.9740	.2624			148			148	.138	.3627		86			86	.1278	.9198	
159			.9969	.0000			144			144	.9960	.0905		17			17	.9886	.9481	
128			.9906	.1731			169			169	.9742	.2527		106			101	.3806	.9120	
32			.9969	.0000			53			53	.9826	.1588		117			113	.3806	.9120	
73			.9969	.0000			200			200	.9826	.1588		117			117	.97	.97	
142			.1000	.0203			130			130	.9999	.0115		111			120	.3209	.9759	
149			.8417	.5939			29			29	.8200	.5273		118			118	.0749	.9165	
66			.8794	.4671			22			22	.9878	.1857		349			403	.2446	.9966	
38			.9966	.0288			155			158	.9365	.3056		136			136	.0850	.9694	
57			.9988	.0847			39			39	.9993	.0731		20			20	.7640	.6582	
96			.9942	.1702			130			130	.9077	.4016		71			65	.7996	.6005	
128			.9776	.2015			22			24	.9914	.1039		250			256	.0325	.9995	
291			.9294	.3961			150			150	.9907	.1631		199			199	.6749	.7837	
260			.9945	.1404			89			89	.9807	.3457		131			124	.3086	.9152	
190			.9999	.0014			162			151	.9991	.0143		47			47	.9910	.9910	
271			.9999	.0000			153			147	.9878	.1857		222			222	.222	.9999	
272			.8004	.5999			107			91	.9363	.3153		225			238	.1621	.9688	
193			.9701	.1608			133			126	.9538	.3005		186			186	.0289	.9759	
163			.9943	.1608			133			126	.9538	.3005		186			186	.0289	.9759	
340			.9925	.3870			185			185	.9935	.1410		95			95	.87	.87	
154			.9925	.3870			185			185	.9935	.1410		95			95	.87	.87	
154			.9708	.2938			76			75	.9946	.1308		109			109	.158	.994	
230			.9925	.2827			51			51	.9721	.2436		88			88	.62	.999	
90			.9360	.3250			66			70	.9815	.1197		10			10	.88	.88	
170			.9840	.1404			139			138	.9732	.2027		103			113	.2303	.9371	
86			.9794	.6704			39			39	.9999	.0000		66			66	.58	.9109	
72			.8952	.4546			96			93	.9399	.3144		110			110	.03	.9109	
66			.8421	.6516			83			83	.9475	.0070		209			212	.3105	.9056	
86			.8483	.5925			74			74	.9248	.2836		159			159	.1176	.9591	
132			.9663	.2753			92			92	.9711	.2027		84			84	.9521	.9507	
197			.0015	1.0000			85			85	.9937	.0283		159			159	.1540	.9881	
24			.8109	.5584			91			91	.9937	.0283		159			159	.1540	.9881	
134			.9881	.1448			122			122	.8150	.5975		85			85	.1387	.9093	
193			.9919	.1723			45			45	.9688	.2749		201			201	.192	.9646	
39			.9291	.3969			123			123	.291	.9995	.0032	87			87	.126	.9998	
257			.9653	.0388			201			201	.9999	.0000		140			140	.0398	.9992	
48			.8074	.9887			203			203	.9999	.0000		251			251	.0289	.9759	
53			.9123	.4905			227			227	.2387	.9271		177			177	.173	.9728	
201			.9875	.1343			227			227	.2387	.9271		177			177	.173	.9728	
322			.9875	.1343			227			227	.2387	.9271		177			177	.173	.9728	
207			.9733	.2926			289			274	.6261	.7977		81			76	.2078	.9872	
219			.9803	.1698			149			149	.3730	.9728		68			72	.9990	.9545	
119			.9759	.1816			186			179	.4744	.8083		161			161	.151	.908	
293			.9883	.1254			186			179	.4744	.8083		161			161	.151	.908	
308			.9883	.1254			186			179	.4744	.8083		161			161	.151	.908	
202			.9763	.2613			111			111	.8048	.7957		87			87	.1857	.9286	
393			.9763	.2613			111			111	.8048	.7957		87			87	.1857	.9286	
30			.5641	.8527			122			133	.1187	.9299		66			90	.4552	.8094	
163			.9960	.0985			219			219	.4092	.9214		66			66	.90	.9794	
192			.9999	.0512			369			361	.3571	.9431		10			10	.83	.83	
96			.0038	1.0000			198			198	.2847	.9856		184			184	.184	.9265	
134			.9996	.0728			281			281	.9996	.0728		192			192	.1072	.9496	
147			.9483	.3713			297			297	.1543	.9880		152			152	.87	.9254	
120			.9600	.2081			353			353	.344	.9880		152			152	.87	.9254	
30			.9600	.2081			353			353	.344	.9880		152			152	.87	.9254	
136			.9858	.1768			227			206	.6675	.7446		96			86	.1000	.0007	
64			.9903	.1343			301			288	.888	.1000		112			112	.195	.9625	
26			.9874	.2957			218			211	.888	.1000		112			112	.195	.9625	
361			.9998	.0913			172			172	.172	.9999		185			187	.8814	.8675	
211			.9696	.2448			172			170	.0108	.9999		100			99	.5241	.8157	
291			.9345	.3559			160			160	.0584	.9893		95			102	.0225	.9163	
209			.9961	.0787			156			144	.3168	.9845		95			95	.0225	.9163	
69			.9924	.0999			85			85	.0488	.9908		13			21	.2334	.9274	
208			.9950	.0999			85			85	.0488	.9908		13			21	.2334	.9274	
263			.9950	.0999			85			85	.0488	.9908		13			21	.2334	.9274	
80			.9950	.0999			85			85	.0488	.9908		13			21	.2334	.9274	
32			.1061	.9493			124			124	.0500	.9815		146			146	.0279	.9996	
153			.9785	.2164			49			49	.1721	.9581		23			23	.8822	.9421	
206			.9828	.1485			250			222	.2047	.9878		153			164	.1031	.9497	
176			.9828	.1485			250			222	.2047	.9878		153			164	.1031	.9497	
141			.9993	.0831			100			114	.9994	.0538		144			136	.4199	.9706	
158			.9825	.1484			280			279	.3301	.9440		149			167	.5204	.8359	
125			.9825	.1484			280			279	.3301	.9440		149			167	.5204	.8359	
110			.9890	.1749			335			311	.0339	.9840		133			130	.1227	.9044	
94			.9966	.0285			110			102	.2096	.9778		19			29	.6238	.7186	
108			.9966	.0285			110			102	.2096	.9778		19			29	.6238	.7186	
169			.9187	.3499			213			213	.2193	.9775		104			104	.0603	1.0000	
178			.9740	.2624			2													

Discussion

The main features of the structure are as described by Lippert & Truter (1960). Differences in detail are primarily due to errors in their determination* To facilitate comparison we have adopted their notation for the several atoms (Fig. 1).

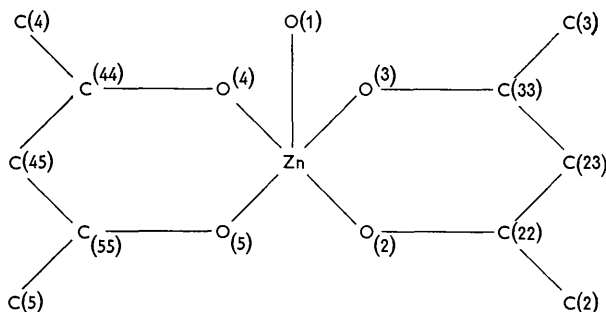


Fig. 1. Numbering of the atoms.

The bond lengths and angles, with their estimated standard deviations (Jeffrey & Cruickshank, 1953) are given in Table 4.

The five oxygen atoms coordinated to the zinc ion appear to be all at the same distance, 2.02 ± 0.02 Å. The arrangement of the oxygen atoms is intermediate between tetragonal pyramidal and trigonal bipyramidal, although somewhat nearer to the former. A comparison of the two models is shown by the bond angles in Table 5. The angles for the tetragonal pyramidal model were calculated with Zn-O distances of 2.02 Å and with the Zn atom 0.41 Å above the base plane.

The mean deviation of the bond angles is 2.9° from the tetragonal pyramid and 10.5° from the trigonal bipyramid. If a bipyramidal model is used in which the trigonal symmetry is not required (*i.e.* symmetry C_s rather than C_{3h}) the mean deviation of the 7 symmetry-fixed angles is 6.4° , while the mean deviation of all angles (taking the deviations in the equatorial plane to be zero) is 4.5° . The most obvious and important deviation of the molecule from the

* Note by Mary R. Truter and E. L. Lippert: We are grateful to the authors of this paper for pointing out that in Lippert & Truter (1960), the angle β is quoted as 93.6° whereas the structure factors quoted correspond to $\beta = 86.4^\circ$. The refinement has been repeated with this error corrected and an empirical extinction correction has also been applied; the fractional coordinates do not differ significantly from those published and only for two values of y (for C(3) and C(44)) is the difference between our corrected results and those in this paper greater than twice the standard deviation. The effect of the correction of β (and a change in b_0 to 5.376 Å by a new Straumanis determination) is to alter the bond lengths round the zinc atom significantly but neither the general stereochemistry of the molecule nor the dimensions within the β -diketone have been changed significantly. Revised values of the Zn-O bond lengths are 2.00-2.02 Å for all bonds including that to the water molecule. The arrangement of the five ligand atoms about the zinc is essentially unchanged, that is it can be regarded as a distorted trigonal bipyramid or as a distorted tetragonal pyramid.

Table 4. Bond lengths and angles in zinc acetylacetonate monohydrate

Estimated standard deviation in parentheses			
Bond lengths (Å)		Bond angles ($^\circ$)	
Zn-O(1)	1.999 (0.02)	O(1)-Zn-O(2)	104.9 (0.8)
Zn-O(2)	2.026 (0.02)	O(1)-Zn-O(3)	97.5 (0.8)
Zn-O(3)	2.038 (0.02)	O(1)-Zn-O(4)	104.9 (0.8)
Zn-O(4)	2.005 (0.02)	O(1)-Zn-O(5)	100.3 (0.8)
Zn-O(5)	2.011 (0.02)	O(2)-Zn-O(3)	88.5 (0.8)
O(2)-C(22)	1.303 (0.048)	O(4)-Zn-O(5)	88.0 (0.8)
O(3)-C(33)	1.290 (0.048)	Zn-O(2)-C(22)	127.3 (1.3)
O(4)-C(44)	1.270 (0.048)	Zn-O(3)-C(33)	126.5 (1.3)
O(5)-C(55)	1.301 (0.048)	Zn-O(4)-C(44)	129.3 (1.3)
C(22)-C(23)	1.404 (0.058)	Zn-O(5)-C(55)	125.2 (1.3)
C(33)-C(23)	1.373 (0.058)	O(2)-C(22)-C(23)	123.3 (1.8)
C(44)-C(45)	1.404 (0.058)	O(3)-C(33)-C(23)	125.5 (1.8)
C(55)-C(45)	1.400 (0.058)	O(4)-C(44)-C(45)	122.7 (1.8)
C(2)-C(22)	1.488 (0.064)	O(5)-C(55)-C(45)	125.4 (1.8)
C(3)-C(33)	1.487 (0.064)	C(22)-C(23)-C(33)	126.8 (2.1)
C(4)-C(44)	1.508 (0.064)	C(44)-C(45)-C(55)	125.1 (2.1)
C(5)-C(55)	1.556 (0.064)	O(2)-C(22)-C(2)	116.7 (2.2)
		O(3)-C(33)-C(3)	115.6 (2.2)
		O(4)-C(44)-C(4)	118.7 (2.2)
		O(5)-C(55)-C(5)	115.2 (2.2)

Table 5. Bond angles in $Zn(acac)_2 \cdot H_2O$

Angle	Actual	Tetr. pyr. model	Diff.	Trig. bipy. model	Diff.
O(1)-Zn-O(2)	104.9	101.7	3.2	120	15.1
O(1)-Zn-O(3)	97.5	101.7	4.2	90	7.5
O(1)-Zn-O(4)	104.9	101.7	3.2	120	15.1
O(1)-Zn-O(5)	100.3	101.7	1.4	90	10.3
O(2)-Zn-O(3)	88.5	87.8	0.7	90	1.5
O(2)-Zn-O(4)	150.1	156.7	6.6	120	30.1
O(2)-Zn-O(5)	85.0	87.8	2.8	90	5.0
O(3)-Zn-O(4)	89.4	87.8	1.6	90	0.6
O(3)-Zn-O(5)	162.0	156.7	5.3	180	18.0
O(4)-Zn-O(5)	88.0	87.8	0.2	90	2.0

tetragonal pyramidal model is the deviation of the basal set of four O atoms from planarity, atoms O(3) and O(4) lying 0.1 Å above and O(2) and O(5) lying 0.1 Å below their mean plane.

The two acetylacetonate groups are nearly planar, but not coplanar, each group being tilted about 12° from normality to the Zn-H₂O bond.

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